

اللَّهُ سَتِّعِينِ  
وَبِيَدِهِ  
الْأَرْحَمِينَ الرَّحْمَنُ الرَّحِيمُ





**Title:**

**Study of specific migration of Bisphenol A from plastic packaging into food stimulants by HPLC**

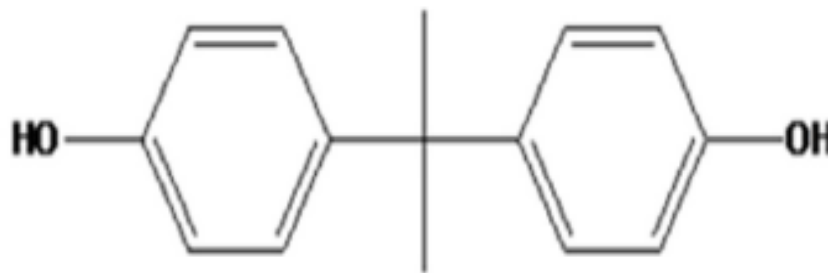
**By:**

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# What is Bisphenol A?

- Bisphenol A, also known as BPA, is a commonly used **chemical** to make polycarbonate plastics.
- Before its known use as a chemical that makes plastic, it was discovered as a **synthetic estrogen** in the 1930's.



# Who discovered its uses?

- The first reported synthesis of BPA was done by Thomas Zincke at the University of Marburg, Germany.
- Dodds and Lawson first showed BPA to be **oestrogenic** in ovariectomized **rats** in 1938.
- Then in 1953 Dr. Herman Schnell of Bayer, Germany developed **manufacturing** processes for a new plastic, **polycarbonate**, that used BPA as a starting material.
- In the same years large-scale uses for polycarbonate plastics.

# Where is it in the environment?

- There are many places where BPA is found:
  - Baby bottles and water bottles
  - Dental sealants and medical equipment
  - As coatings on soda cans
  - Household appliances, CD's & DVD's





Resin  
identification  
code



# Why is BPA used so much?

- many things made of plastic are made with BPA
- BPA is used so much because it is highly durable
- It has a high heat-resistance
- It is shatter resistance and has high optical clarity.

# Why is there such argument?

- The **plastic industry** is stating that the amount of BPA that a normal person is exposed to is **not toxic**.
- Other **scientists** and researchers have found that their research proves that it causes **cancer** and **premature maturity**.



# The independent researcher's opinion

- BPA binds with the **estrogen receptor** and has the capacity to **stimulate** estrogenic responses.
- This means that it is **biologically active** and therefore not "**totally safe.**"
- Moreover, there are a series of animal studies that show that bisphenol A actually does lead to developmental abnormalities in **laboratory animals** at **low levels** of exposure.
- 
- This also directly **contradicts** their **assertion** that it is totally safe.

## What we can do to minimize exposure...

- Use **filter systems** instead of bottled water
- Buy **fresh** produces instead of canned food
- Choose **baby bottles** and cups made of **glass** or polyethylene (#1,#2,#4 recycling symbols), or polypropylene (#5)
- Store food in glass, ceramic or metal containers.
- Instead of having young children get **dental sealants** for teeth that will fall out within a few years, and have them increase their practice of at **home dental care**.

# Analysis Methods:

- BPA has been determined with the use of GC/MS or HPLC.
- These instrumental methods are **expensive** and need complicated pretreatment and are not applicable to onsite monitoring.
- **Electrochemical techniques** have been recognized as suitable methods for detection of phenolic compounds.

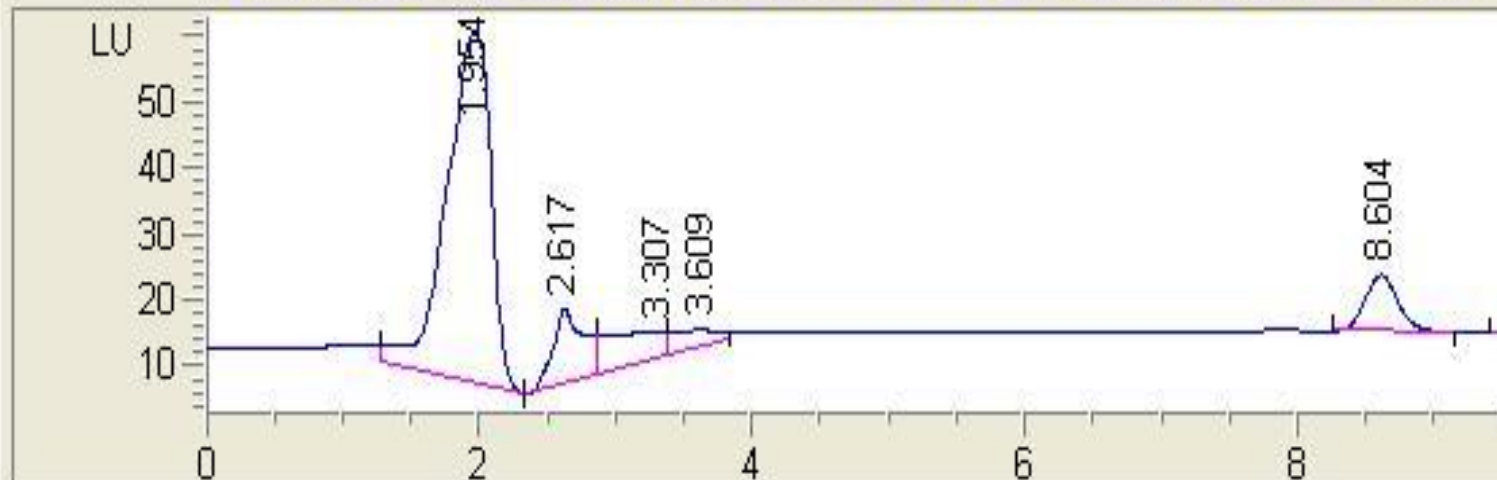
# Analysis Method:

- HPLC(RP)
- Mobile Phase(ACN 40%, H<sub>2</sub>O 60% )
- Flow Rate(1.2 ml/min)
- Detector(FL, Ex=230nm, Em=316nm)
- PMT Gain(10)
- Inj Val(20ul)

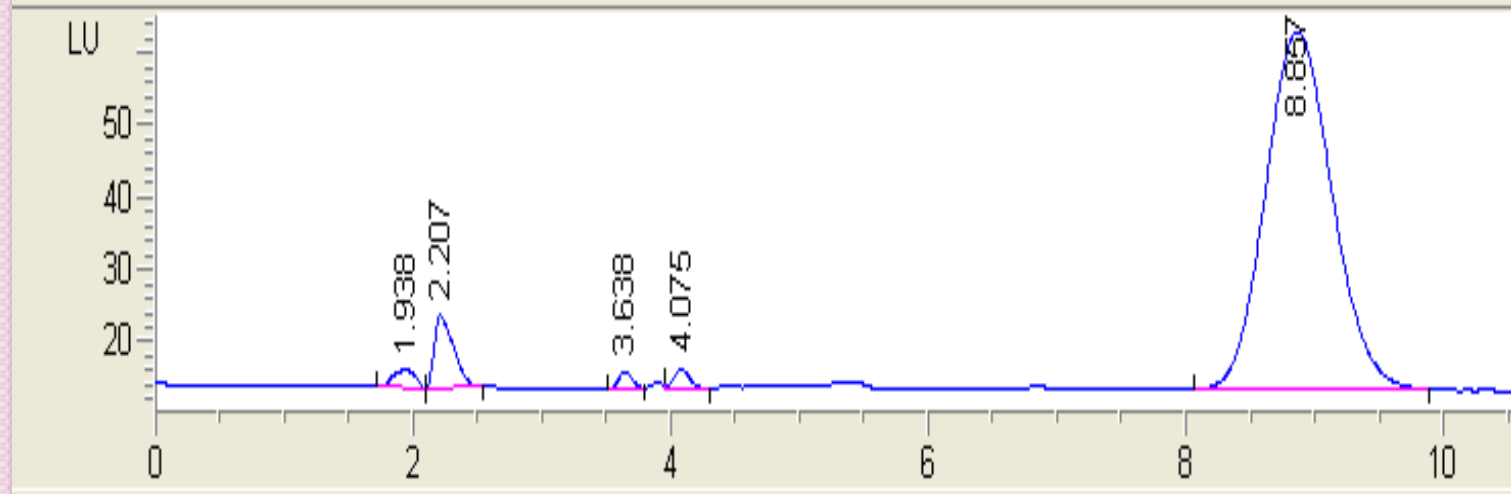


# Chromatogram:

FLD1 A, Ex=230, Em=316 (C:\CHEM32\1\DATA\BPA GANJAL\DEF\_LC 2005-07-

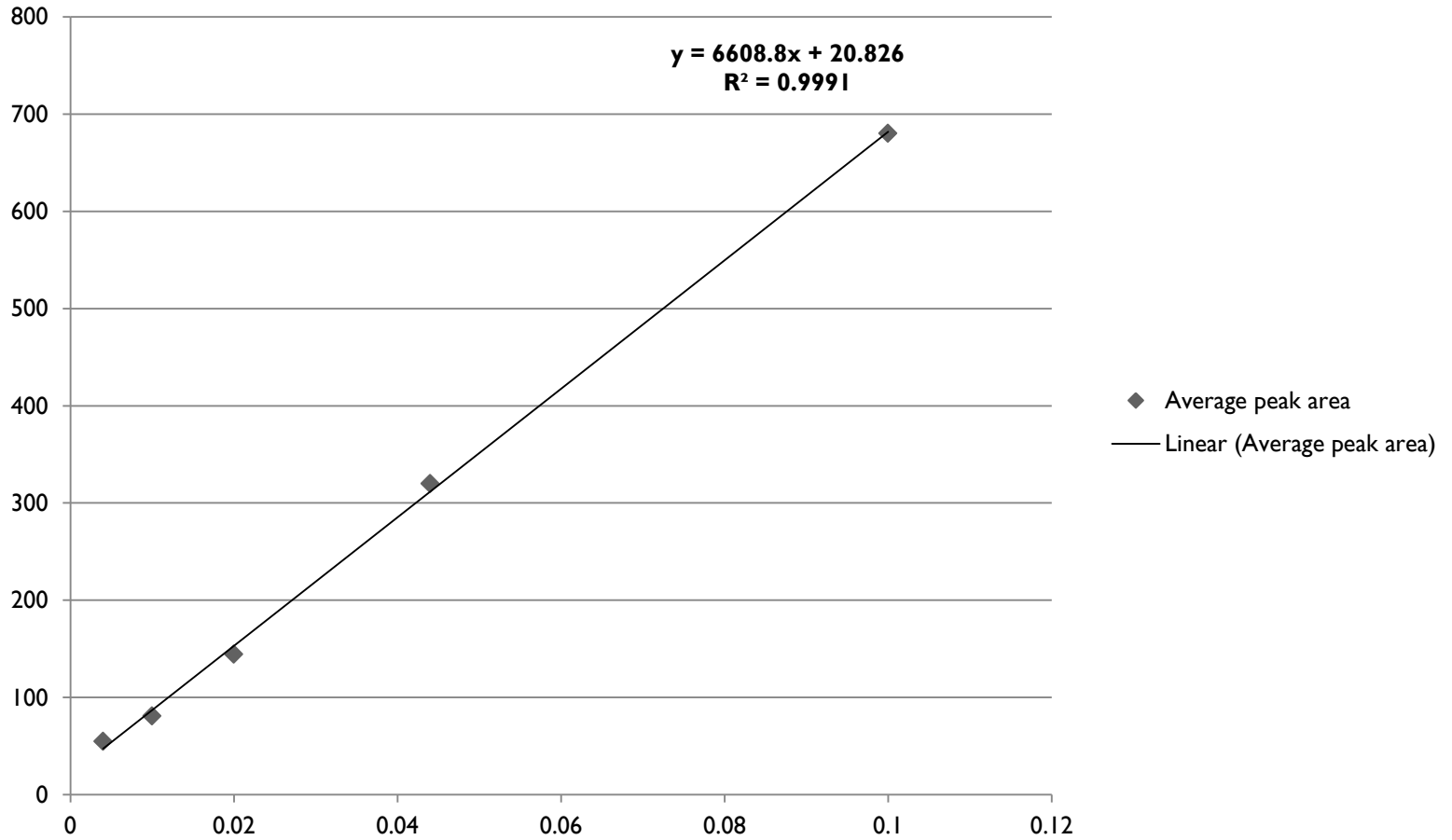


FLD1 A, Ex=230, Em=316 (C:\CHEM32\1\DATA\BPA GANJAL\DEF\_LC 2005-07-29 07-21-14\021-0101.D)



# Calibration Curve:

## Average peak area



# Voltammetry

Voltammetry = Volt-Am(pero)-Metry

Voltage ramp applied to electrode

Current measured

$$I = f(U)$$

Method first described 1922 by Heyrovsky

# Advantages of this method:

1- Repeadability

2- low cost

3- low analysis time

4- low LOD

5- Enviromentally Friendly



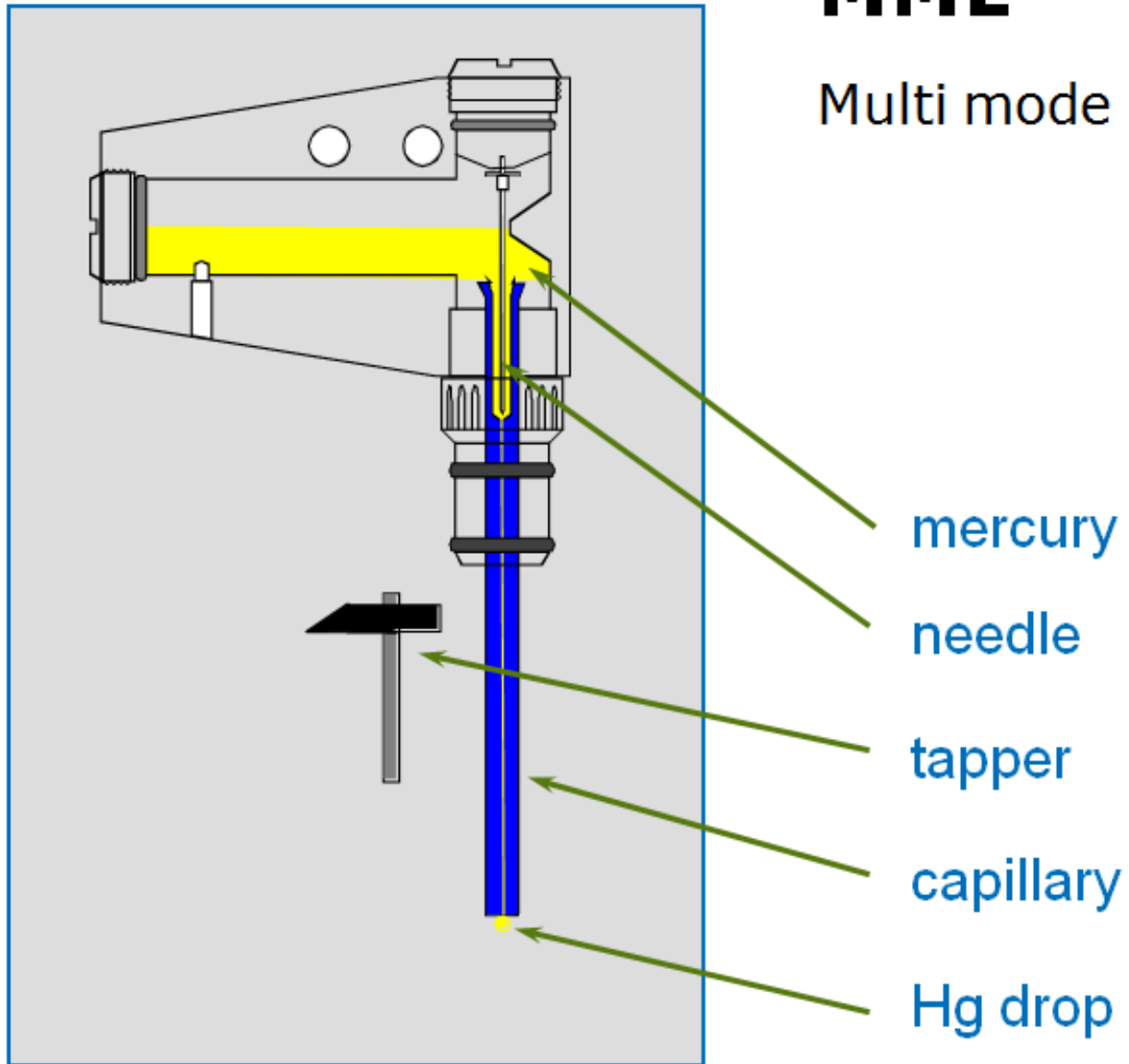
# Sensitivity

<b>Titration</b>	<b>IC</b>	<b>VA</b>
0.000 1% = 1 ppm to 100%	0.000 000 1% = 1 ppb to 0.1%	0.000 000 000 1% = 1 ppt to 0.000 1% = 1 ppm
Major constituents	Anions	Cations

Voltammetry → Traces of heavy metals

# MME

Multi mode electrode



Thanks  
for  
your  
attention

